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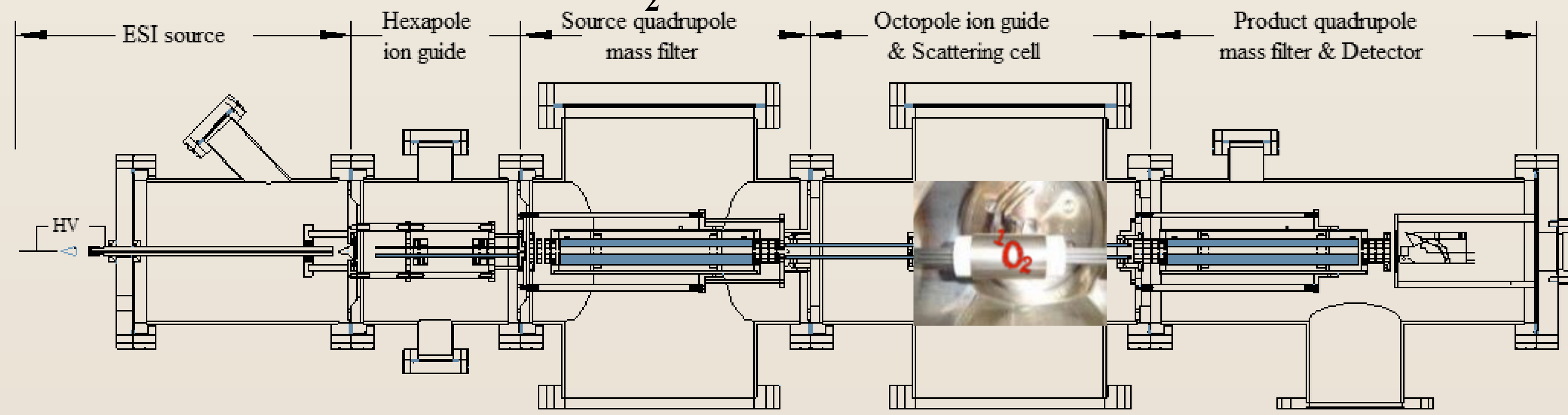
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### Introduction

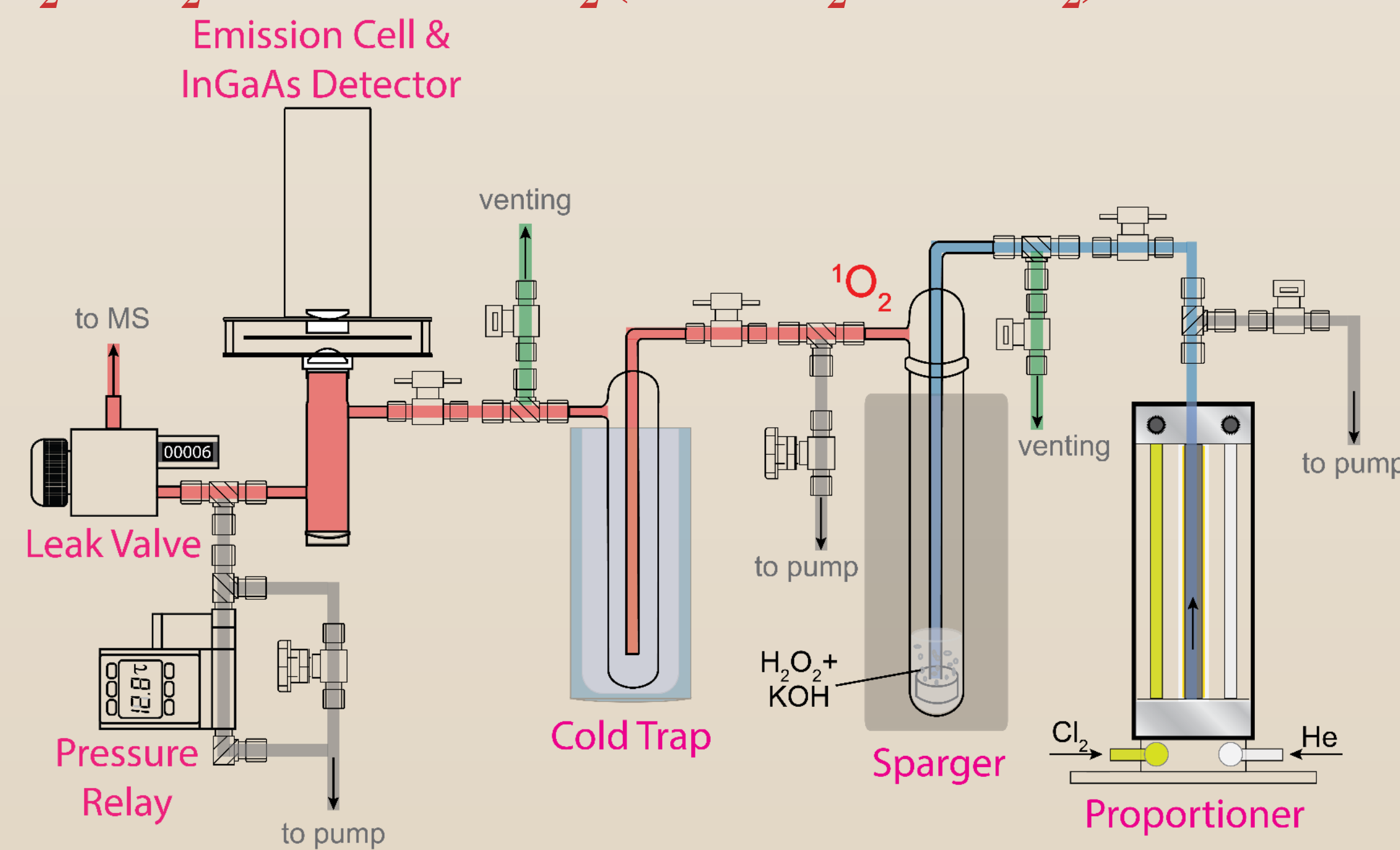
- Guanine, the preferred target in DNA for bromination by reactive halogen species (RHS), forms the mutagenic lesion 8-Bromoguanine (8BrG)
- Biological significance of oxidatively generated radical cations of 8-Bromoguanine (8BrG<sup>•+</sup>) and 8-Bromoguanosine (8BrGuo<sup>•+</sup>) by the lowest electronically excited singlet O<sub>2</sub> (a<sup>1</sup>Δ<sub>g</sub>):
  - ❖ Chronic inflammation and Cancer
  - ❖ DNA strand breaks, DNA-protein crosslinks, mutation and apoptosis
  - ❖ Biomarker for oxidative and halogenative stress within cells and tissues
  - ❖ Singlet O<sub>2</sub>-based photodynamic therapy in cancer treatment
  - ❖ Brominated nucleotides are considered potential radiosensitizers in radiotherapy

### Instrumentation & Experiment

#### I. Guided-Ion-Beam Tandem Mass Spectrometer for Generation of 8BrG<sup>•+</sup> and Reaction with <sup>1</sup>O<sub>2</sub>



#### II. Generation and Detection of <sup>1</sup>O<sub>2</sub> Before Leaking into the Scattering Cell of the Mass Spectrometer



### Theoretical Methods

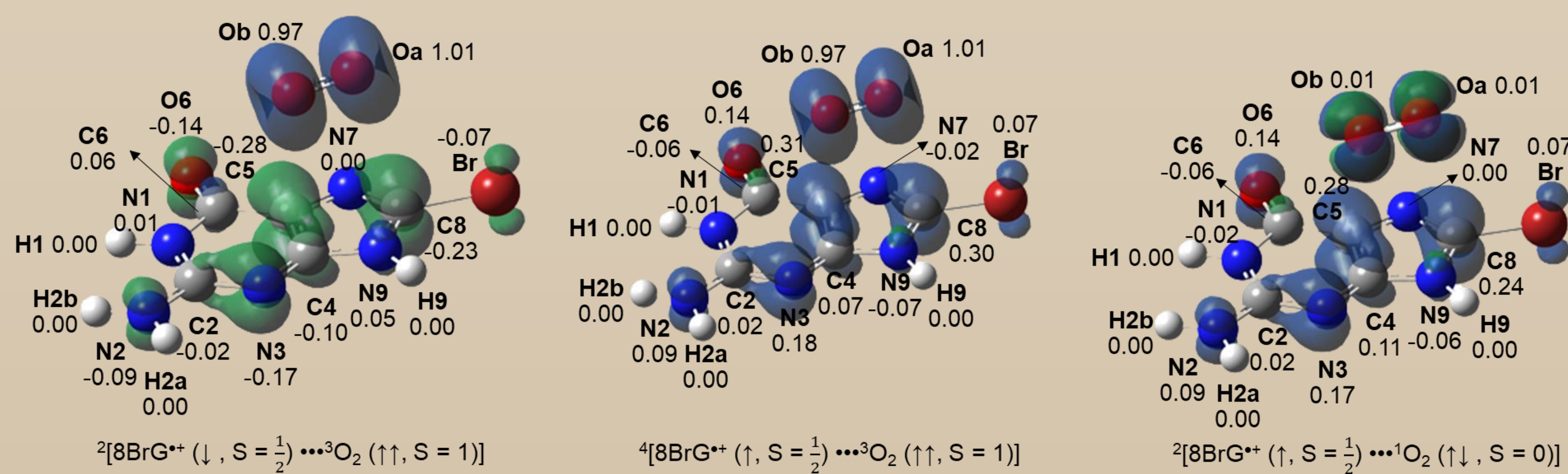
The calculation of the reaction potential energy surface for 8BrG<sup>•+</sup> with <sup>1</sup>O<sub>2</sub> is challenging due to multiconfigurational wavefunctions originating from the mixed open- and closed-shell character of <sup>1</sup>O<sub>2</sub>, and spin contamination in transition states and products

#### I. Approximately Spin Projected DFT

$$E = \frac{\langle \hat{S}^2 \rangle^{\text{HS}} - \langle \hat{S}^2 \rangle^{\text{BS}}_{\text{exact}}}{\langle \hat{S}^2 \rangle^{\text{HS}} - \langle \hat{S}^2 \rangle^{\text{BS}}} E^{\text{BS}} - \frac{\langle \hat{S}^2 \rangle^{\text{BS}} - \langle \hat{S}^2 \rangle^{\text{BS}}_{\text{exact}}}{\langle \hat{S}^2 \rangle^{\text{HS}} - \langle \hat{S}^2 \rangle^{\text{BS}}} E^{\text{HS}}$$

$E^{\text{BS}}$ : energy for broken-symmetry state  
 $\langle \hat{S}^2 \rangle^{\text{BS}}$ : spin angular momentum for broken-symmetry state  
 $E^{\text{HS}}$ : energy for high-spin state  
 $\langle \hat{S}^2 \rangle^{\text{HS}}$ : spin angular momentum operator for high-spin state  
 $N^\alpha$ : number of alpha electrons  
 $N^\beta$ : number of beta electrons

$$\langle \hat{S}^2 \rangle^{\text{BS}}_{\text{exact}} = \frac{N^\alpha - N^\beta}{2} \left( \frac{N^\alpha - N^\beta}{2} + 1 \right)$$



#### II. Coupled-Cluster Theory

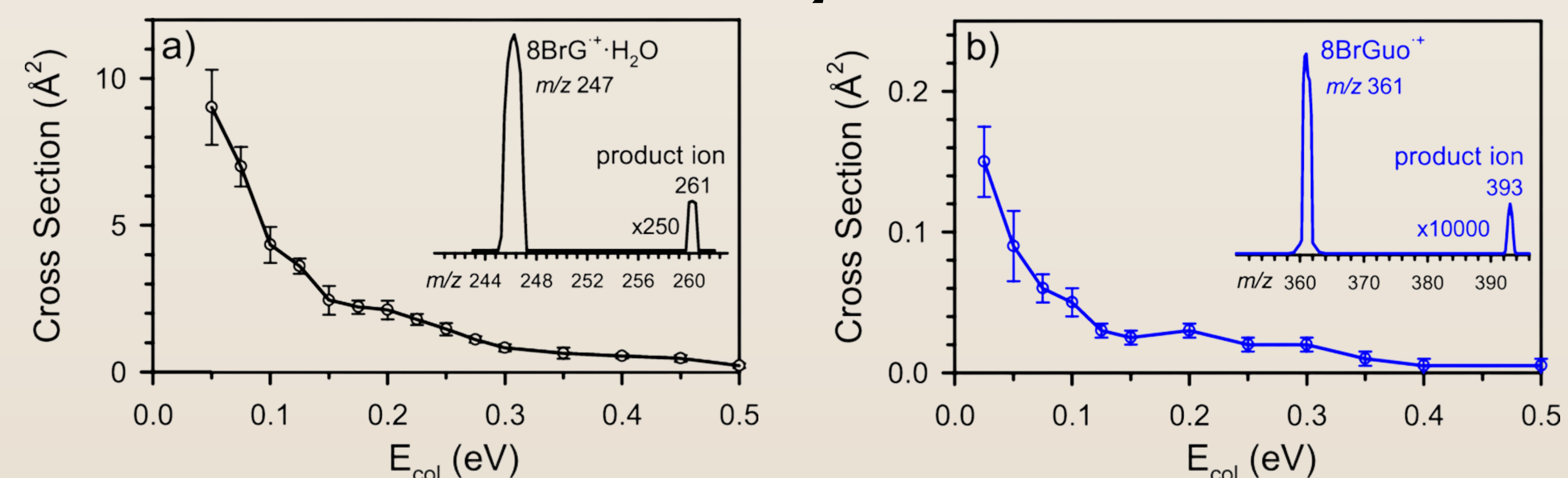
DLPNO-CCSD(T) coupled with the aug-cc-pVTZ basis set was employed to assess the spin contamination in the reaction structures using the T1 diagnostic; it was also considered the gold standard of quantum chemistry due to its accuracy comparable to experiments (in the absence of spin contamination)

#### III. CASPT2

Energies of the DFT optimized reaction structures were recalculated using the multireference active space self-consistent field method CASPT2/6-31G\*\*. It adds dynamic correlation to the complete active space self-consistent field (CASSCF) wavefunction using the second order perturbation theory

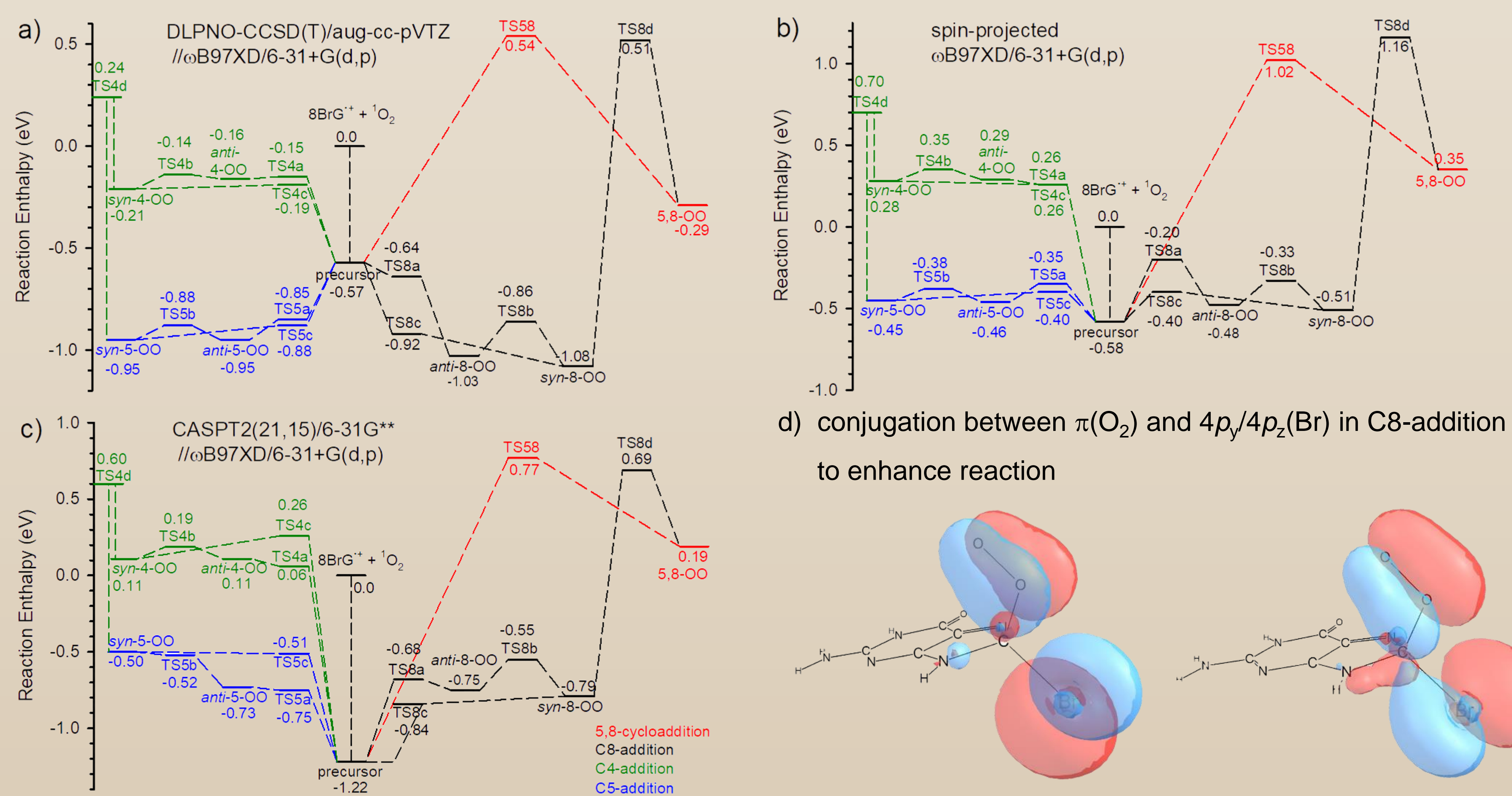
### Results

#### I. Reaction Products and Cross Sections for <sup>1</sup>O<sub>2</sub> with 8BrG<sup>•+</sup> and 8BrGuo<sup>•+</sup>



- Reaction is exothermic and has no activation barriers above starting reactants
- The reactivity of 8BrG<sup>•+</sup> is 5-fold higher than that of the unsubstituted guanine (G<sup>•+</sup>)
  - The reaction efficiency of 8BrG<sup>•+</sup> is 9.5% at  $E_{\text{col}} = 0.05$  eV and decreases to 3% at 0.4 eV
  - The reaction efficiency of G<sup>•+</sup> is 2% at  $E_{\text{col}} = 0.05$  eV and decreases to 1.4% at 0.1 eV

#### II. Reaction Potential Energy Surface of <sup>1</sup>O<sub>2</sub> with 8BrG<sup>•+</sup> at different levels of theory



Among the different computational methods used in the PES calculations, the CASPT2(21,15) theory provided the best description and agreement with experimental data

### Conclusions

- In the reaction of <sup>1</sup>O<sub>2</sub> with 8BrG<sup>•+</sup>/8BrGuo<sup>•+</sup>, the product cross section increases with decreasing  $E_{\text{col}}$ .
- It indicates that the reaction is exothermic
- The most energetically favorable reaction pathways for 8BrG<sup>•+</sup> + <sup>1</sup>O<sub>2</sub> are the C5-O<sub>2</sub> and C8-O<sub>2</sub>

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### References

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